



## Engine Combustion Test using Algae Biofuel with Nanofluid

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The research attempts to analyze the influence of *S. Marginatum* algae biodiesel, used in a diesel engine with additives. Transesterification was used to prepare biodiesel from microalgae to mix with standard diesel at a rate of 20% (vol.) called B20. The nanofluid  $Al_2O_3$  was prepared and tested along with B20 under different loads at a speed of 1500 rpm in a direct ignition diesel engine for their combustion characteristics. It is concluded that fuel mixtures of  $Al_2O_3$  nanofluids confirmed better combustion on a CI engine compared to B20 mix.

**Keywords:** Algae biomass,  $Al_2O_3$  nanoadditive, Biodiesel, Blending, Combustion

### Introduction

Renewable fuels add more to the fossil fuel reserves. Biodiesel will be used as a synthetic fuel to meet current and future energy demands. Compared to diesel, biodiesel emits low carbon and smoke emissions which decreases climate change.<sup>1</sup> Studies have indicated that lower heating value and high emissions of  $NO_x$  restrict biodiesel usage in the diesel engine. Earlier studies suggested that biodiesel mixed with nanoparticles could boost engine efficiency and at the same time leading to a significant decrease in engine exhaust emissions. Earlier studies showed that adding biodiesel additives would boost engine efficiency while reducing  $NO_x$  emissions but increasing gradually other emissions.<sup>2</sup> Earlier studies have illustrated that biodiesel mixtures with additives and nanoparticles found more benefits than their tidy form. A few studies have addressed the effects of biodiesel blends with various additives on a CI engine. The current study examines the results of biodiesel blends on a CI engine with  $Al_2O_3$  nanofluids applied to diesel. The research work focuses on the impact of a CI engine's combustion parameters.

### Materials and Methods

#### Preparation of Biodiesel and Nanofluids

Alkaline transesterification technique was used to produce SMME (*S. Marginatum* Methyl Ester) by taking the raw algae oil. Methoxide solution mixing

is the primary step in SMME production using ultrasonicator. It is then heated with crude algae oil for around 60 minutes up to 65°C. It was left for settling and separation of glycerol layer in 24 hours happens due to its higher density. Upper layer is crude biodiesel. The crude biodiesel is then washed with distilled water until the wastewater shows no colour change with phenolphthalein indicator. Crude biodiesel was then kept overnight with anhydrous sodium sulfate for dehydrating the residual water content. Blending of biodiesel along with  $Al_2O_3$  nanofluids additives was done using ultrasonicator.<sup>3</sup> Three fuels were prepared and tested. One is B20 formed by 20 % blending of algal biodiesel with petro-diesel. The other two are addition of 0.1 and 0.2% (vol./vol.) of  $Al_2O_3$  nanofluid to B20. For several days the formulated nanofluids have remained highly stable. ASTM specifications are used to assess the properties of test fuel blends as shown in Table 1. The specifications of nanofluids are shown in Table 2. There is no particulate sedimentation in the prepared nanofluids.

#### Nanoparticle Size Analysis using XRD

XRD analysis was performed with X'pert powder XRD method, PANalytical X-Ray diffractometer with  $Co-K\alpha$  radiation in the range of 10–90°, to analyse the crystal structure, size and purity of the purchased  $Al_2O_3$  nanoparticles. To assess the crystalline size an X-Ray beam was passed through a nanoparticle sample with a wavelength of 0.15 nm. The XRD patterns of the purchased  $Al_2O_3$  nanoparticles specifically are shown in Fig. 1. All peaks in XRD

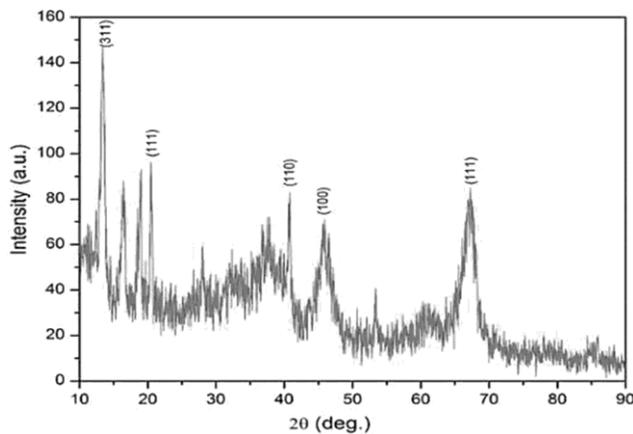
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Table 1 — Physio-chemical properties and details of Al<sub>2</sub>O<sub>3</sub> nanoparticle

| Properties                                       | ASTM D975 | ASTM 6751 | Diesel | B100  | B20   | B20+0.1Al <sub>2</sub> O <sub>3</sub> | B20+0.2Al <sub>2</sub> O <sub>3</sub> |
|--|-----------|-----------|--------|-------|-------|---------------------------------------|---------------------------------------|
| Density at 15 °C (kg/m <sup>3</sup> )            | —         | 861–901   | 851    | 891   | 989   | 986                                   | 986                                   |
| Kinematic viscosity at 40°C (mm <sup>2</sup> /s) | 2–4.2     | 3.6–5.1   | 2.7    | 4.85  | 3.3   | 3                                     | 3.1                                   |
| Calorific value(kJ/kg)                           | —         | —         | 44841  | 42053 | 41204 | 45364                                 | 45529                                 |
| Cetane number                                    | 41        | 48        | 47     | 64    | 53    | 52                                    | 54                                    |
| Flash point(°C)                                  | 53        | 102       | 65     | 129   | 98    | 83                                    | 83                                    |
| Fire point(°C)                                   | —         | —         | 71     | 137   | 111   | 110                                   | 111                                   |

Table 2 — Details of Al<sub>2</sub>O<sub>3</sub> nanoparticle

|                       |                                |
|-----------------------|--------------------------------|
| Company               | Sigma Aldrich                  |
| Specific Surface area | 40 m <sup>2</sup> /g           |
| Number of CAS         | 1344-28-1                      |
| Molecular Weight      | 101.96                         |
| Color                 | White                          |
| Linear Formula        | Al <sub>2</sub> O <sub>3</sub> |
| Size of particle      | <50nm                          |
| MDL number            | MFC00003424                    |

Fig. 1 — XRD patterns for Al<sub>2</sub>O<sub>3</sub> nanoparticle

patterns clearly demonstrate good crystal structure of both nanoparticles and detect no peaks of impurities. The XRD test results therefore show that the purity of purchased nanoparticles exceeds 99%. The Al<sub>2</sub>O<sub>3</sub> nanoparticles obtained sizes, 50 nm, and 27 nm, respectively.<sup>4</sup>

### Experimental Setup

A single-cylinder diesel engine and an eddy current dynamometer with an exhaust gas analyser were included in the analytical test setup. The impact of the real-time sensors is processed with a data acquisition system on the computer. In this analysis the variations in output and exhaust emissions are investigated. Digital tachometer is being used to measure engine rpm. Form K thermocouples are used to measure the temperature of the exhaust gas. The data acquisition

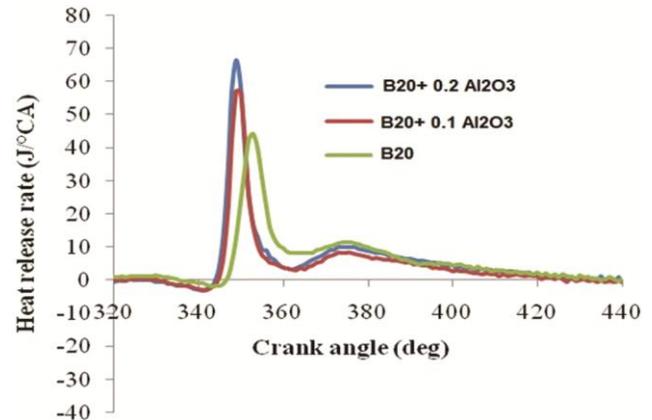


Fig. 2 — Variation in heat release rate with changing crank angle method based on LabVIEW is used to track the data in real time.<sup>5</sup>

## Results and Discussion

### Heat Release Rate (HRR)

The HRR for Al<sub>2</sub>O<sub>3</sub> nanofluids B20, 0.1 and 0.2 at 100% load is shown in Fig. 2. Premixed phase for Al<sub>2</sub>O<sub>3</sub> nanofluids is found to be higher than B20. Owing to the high viscosity of the B20 less fuel is being required for the premixed phase and most of the combustion occurs in the diffusion combustion process. More diffusion oxidation is also led to elevated smoke emissions. Higher HRR in premixed combustion is observed with Al<sub>2</sub>O<sub>3</sub> nanofluids as compared to B20 due to lower biodiesel viscosity. In the premixed process, higher HRR of Al<sub>2</sub>O<sub>3</sub> nanofluids results in greater peak pressure.<sup>6</sup>

### Cylinder Pressure (CP)

The cylinder pressure variation at 100% load according to the angle of the crank is shown in Fig. 3. For B20, lesser air-fuel mixture is formed during the pre-mixed combustion process that leads to lower heat release and therefore lower peak pressure. During the diffusion combustion process, the heat release is delayed and it produces even more heat, which leads to lower peak pressure, higher EGT and reduced energy transfer to useful work. With processing of Al<sub>2</sub>O<sub>3</sub> nanofluids,

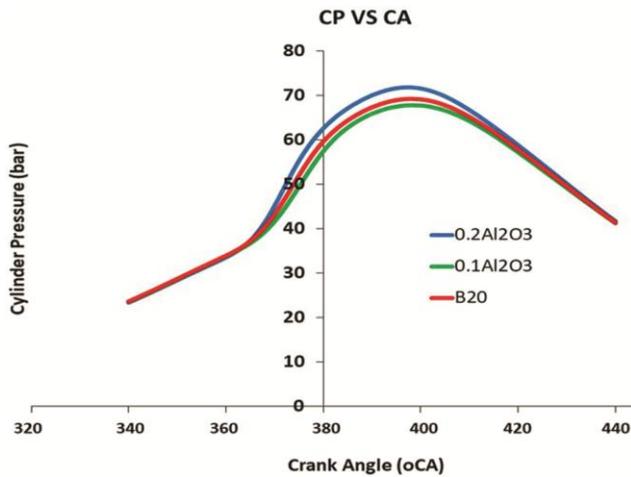


Fig. 3 — Variation of cylinder pressure with varying crank angle

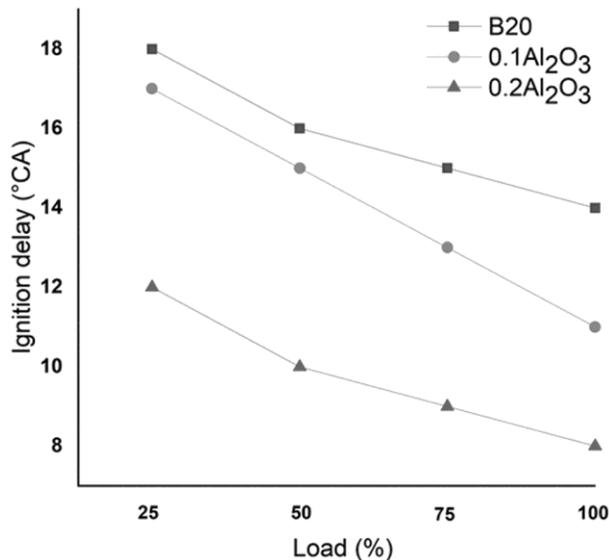


Fig. 4 — Variation of ignition delay with varying load

ignition improved and thus high thermal release occurred in which cylinder peak pressure induced proximity to operation of  $\text{Al}_2\text{O}_3$  nanofluids.<sup>7</sup>

#### Ignition Delay (ID)

The difference in load and ignition delays is shown in Fig. 4. The B20 shows prolonged delays in the ignition compared to nanofluid  $\text{Al}_2\text{O}_3$  pre-mix, due to decreased volatility, high viscosity and density. This leads in inadequate atomization as well as vaporisation. Ignition delay decreased at all charges with B20 leading to enhanced fuel properties like lower viscosity that reduces the delay in physical ignition. Ignition delay for  $\text{Al}_2\text{O}_3$  nanofluid pre-mix and B20 is  $12^\circ\text{CA}$ ,  $11^\circ\text{CA}$ , and  $9^\circ\text{CA}$  at a load of 100%, respectively.<sup>8</sup>

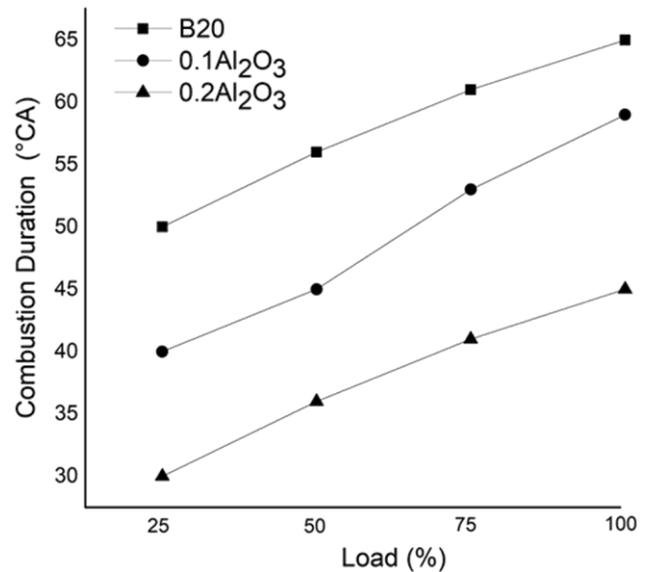


Fig. 5 — Variation of combustion duration with varying load

#### Combustion Duration (CD)

The difference in the time of combustion with load for the different test fuels are shown in Fig. 5. Due to poor fuel spray characteristics of B20, Combustion duration (CD) is longer indicating inferior combustion. Combustion duration for B20 and  $\text{Al}_2\text{O}_3$  nanofluid pre-mix is  $67^\circ\text{CA}$ ,  $58^\circ\text{CA}$ , and  $47^\circ\text{CA}$  respectively, at 100% load. Combustion duration is lower for  $\text{Al}_2\text{O}_3$  nanofluid pre-mix in comparison to B20. This is due to better combustion of  $\text{Al}_2\text{O}_3$  nanofluids as a result of modified fuel properties such as lower density and viscosity.

#### Conclusion

In the present study, the combustion characteristics of a single-cylinder engine running on B20 blend of *S. Marginatum* algae, and its pre-mix with 0.1 and 0.2  $\text{Al}_2\text{O}_3$  nanofluid were investigated. The effects of various parameters, such as HRR, cylinder pressure, percentage mass burnt, delay in ignition and duration of combustion was studied. Adding  $\text{Al}_2\text{O}_3$  nanoparticles to B20 fuel and running the engine at constant speed and constant variable load indicated enhancements in engine combustion as load power increased and the ignition delay as well as the combustion duration decreased.

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